

Carbon-Negative Hydrogen: Aqueous Phase Reforming (APR) of Glycerol over NiPt Bimetallic Catalyst Coupled with CO₂ Sequestration

Background/Objective

Hydrogen (H₂) is an important chemical in refining processes, ammonia production, energy storage, and as an alternative and cleaner fuel. Biomass captures CO₂ from the atmosphere during the process of photosynthesis. Using biomass to produce H₂ offers the advantage of producing H₂ from a low-carbon source. Combining H₂ production from biomass with carbon capture and storage during H₂ production has the potential to produce carbon-negative H₂. Herein we report the production of high-pressure, carbon-negative H₂ from glycerol.

Approach

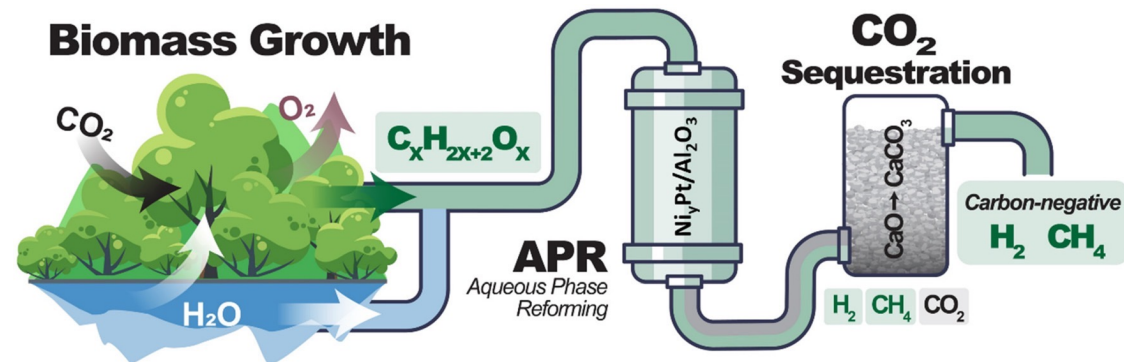
Aqueous phase reforming (APR) of 10 wt% glycerol solution was studied with a series of Nickel Platinum (NiPt) alumina bi-metallic catalysts supported on alumina. The adsorption capacity of the CaO adsorbent was measured at different temperatures.

Results

The Ni₈Pt₁-450 catalyst had the highest hydrogen selectivity (95.6%) and the lowest alkanes selectivity (3.7%) of the tested catalysts. The hydrogen selectivity decreased in the order of Ni₈Pt₁-450 > Ni₈Pt₁-260 > Ni₁Pt₁-260 > Pt-260. Life cycle analysis showed that the APR of glycerol coupled with CO₂ capture has net negative CO₂ equivalent greenhouse gas emissions of -9.9 kg CO₂ eq./kg H₂ and -50.1 kg CO₂ eq./kg H₂ when grid electricity and renewable electricity are used, respectively, and the CO₂ is allocated respectively to the mass of products produced. The cost of this H₂ was estimated to be 2.4 USD per kg H₂ when grid electricity is used and 2.7 USD per kg H₂ when using renewable electricity. The cost of glycerol has the highest contribution of 1.71 USD per kg H₂.

Significance/Impacts

This study shows that combining H₂ production from biomass with carbon capture and storage during H₂ production has the potential to produce carbon-negative H₂.



Schematic representation of a process to produce carbon-negative H₂ from biomass-derived compounds through APR technology and CO₂ sequestration.